

DP-003-1104012

Seat No.

M. Sc. (Sem. IV) (CBCS) Examination

March / April - 2022

Chemistry: C(PA)-404

(Applied Analytical Chemistry) (Ele. - 1) (New Course)

Faculty Code: 003

Subject Code: 1104012

Time : $2\frac{1}{2}$ Hours]

[Total Marks: 70

Instructions: (1) All questions are compulsory.

- (2) All questions carry equal marks.
- 1 Answer the following: (any seven)

14

- (a) What is solvent extraction? Give its separation methods and purpose.
- (b) What are food additives? Give the purpose of its addition.
- (c) How will you collect blood for plasma or whole blood analysis?
- (d) What is solid phase micro extraction?
- (e) Define ore and alloy. What are common alloys?
- (f) In solvent extraction of uranium with 8-hydroxy quinoline in CHCl₃, the volume of aqueous and organic phase were 25 cm³ when the percentage extraction was 99.8%. Calculate the distribution ratio.
- (g) What are food colours? Classify and give their name and role in food preservation.
- (h) How will you estimation of serum albumin and globulin in blood sample ?

	(j)	How will you analyze Al_2O_3 in bauxite sample?	
2	Answer the following: (any two)		14
	(a)	Discuss partition theory of solvent extraction.	
	(b)	Explain solvent extraction by macromolecules in detail.	
	(c)	Give the application of solvent extraction.	
3	Answer the following:		14
	(a)	How will you analyze chloride in blood sample?	
	(b)	Discuss in detail the determination of benzoic acid in food sample.	
		OR	
3	Answer the following:		14
	(a)	How will you analyze CaO and MgO in dolomite?	
	(b)	Discuss microwave enhance chemistry in green analytical chemistry.	
4	Answer the following:		14
	(a)	How will you analyze serum calcium in blood sample?	
	(b)	Write note on composition of blood.	
5	Answer the following: (any two)		14
	(a)	What are emulsifiers, stabilizers and thickeners? Give their names and role of in food materials. How will you detect stabilizing agent in food materials?	
	(b)	How will you analyze copper and zinc in brass sample?	
	(c)	Give the classification of solvent free sample preparation	

Give the principle of green analytical chemistry.

methods in detail.

(i)

(d) Give the analysis of lime-stone in detail.